Compound 20 separated as an oil, which resisted all attempts at crystallization. It was therefore taken up in ether and characterized as the hydrochloride, m.p. 208-210.°

IR spectra: all compounds exhibited a carbonyl

absorption band in the 5.92-5.95 μ region.

3-Substituted 1,5-Diphenylhydantoins (Table III) -A: Bromination of the 4-Imidazolin-2-ones—To a solution of the appropriate 4-imidazolinone (0.01 mole) in chloroform (50 ml.) was slowly added, at room temperature, a solution of bromine (1.9 g., 0.012 mole) in chloroform (10 ml.). The solvent was then distilled (evolution of hydrogen bromide) and the oily residue was crystallized from 95% ethanol to afford the hydantoin.

Compound 14 (R = allyl) absorbed two molar equivalents of bromine to afford in 45% yield 3-(2,3dibromo-1-propyl)1,5-diphenylhydantoin (XI), m.p. 131-133° after crystallization from benzene-petroleum ether. The same compound was obtained on treatment of the allylhydantoin 25 with one molar equivalent of bromine, in 65% yield.

Anal.—Calcd. for C₁₈H₁₆Br₂N₂O₂: C, 47.81; H, 3.56; N, 6.19. Found: C, 47.83; H, 3.73; N, 6.00.

Treatment of the 4-imidazolinones 19, 20 (as the oily base), 21, and 22 with bromine, followed by crystallization of the oily reaction product from ethanol, afforded the hydrobromides of the hydantoins 30 (m.p. 224-226°; N% Calcd. 10.39. Found: 10.30.); 31 (m.p. 189-191°; N% Calcd. 9.72. Found: 9.56.); 32 (m.p. 258-261° dec.; Calcd. 9.72. Found: 9.61)., and 33 (m.p. 287-288° dec.; N% Calcd. 9.41. Found: 9.38.). The salts were converted into the free bases by treatment with 10% sodium carbonate.

B: Treatment of 5-Bromo-3,4-diphenyl-4-oxazolin-2-one (V, R = Ph) with Amines—Mixtures containing 3.16 g. (0.01 mole) of the bromooxazolone (1) and 0.015 mole of the appropriate amine were heated 6 hr. at 100°. Mixtures containing low-boiling amines (n-propyl, n-butyl, isobutyl, and allylamine) were heated at the reflux temperature of the amine for 8-10 hr. The excess amine was then distilled off at reduced pressure and the crude product was crystallized from 95% ethanol.

IR spectra: all compounds exhibited a typical hydantoin carbonyl absorption (6) (two bands in the regions $5.68-5.69 \mu$ and $5.87-5.89 \mu$).

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3, 4 - Diphenyl - 4 - oxazolin - 2 - one derivatives—synthesis

Pharmacological screening—3,4-diphenyl-4oxazolin-2-one derivatives

IR spectrophotometry—structure

Antibacterial and Antifungal Activity of Certain \(\beta\)-Aminoketones

By RAJENDRA S. VARMA* and W. LEWIS NOBLES

Preliminary biological evaluation for 17 compounds is provided. Eleven compounds in this study exhibited some degree of activity.

NDER THE CONDITIONS of the Mannich reaction, a series of β -aminoketones dihydrochlorides was synthesized utilizing 1-(N-β-hydroxyethyl-4-piperidyl)-3-(4-piperidyl)-propane (I) and several aromatic ketones (1). In this report, preliminary screening results for antibacterial and antifungal activities are described.

Several techniques are available to test for antimicrobial activity. Among the in vitro methods are dilution or agar diffusion techniques. The former methods are suitable for assay procedures, but the methods are time consuming for screening of a large number of compounds, and many of them are not satisfactory to determine antifungal activity when filamentous fungi are used as test organisms. This is particularly true if partial inhibition is studied

Received April 17, 1968, from the Department of Pharmaceutical Chemistry, School of Pharmacy, University of Mississippi, University, MS 38677

Accepted for publication June 17, 1968.

This investigation was supported in part by grant No.

This investigation was supported in part by grant No. AI-04701 from the National Institutes of Health, Bethesda, Md. The authors express their deep appreciation for the facilities and assistance provided during this work.

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because it is difficult to determine the amount of growth of these fungi (2). Diffusion methods such as those represented by the use of filter paper disks on an agar plate were chosen because of their suitability for water-soluble compounds and their simplicity of operation.

Table I—Antibacterial and Antifungal Activity of β -Aminoketones

	R—-C—	-СН ₂ -	−CH ₂ −−	-N	(CI	I ₂) ₃ —	N-	−CH₂−	- CH ₂ -	-он -:	2 H Cl		
	Microbial Spectrum ^a												
R		1	2	3	4	5	6	7	8	9	10	11	12
		_	_	_	_	_	_	_	+		_	_	-
Br		-	-	-	_	_	+	+	_		-	_	
Ç _I		+	_	+	+	+		+	+	+	-	-	_
Ç F		-	-	_	_	_	_	_	_	_	_	_	-
CH ₃		_	_	_	_	_	_	-	_	-	_	_	-
CH ₃		-	_	_	+	-	-	+	_	+	-		_
CH ₃		_	_	<u></u>	-	-	_			_		-	-
NO ₂		+	-	+	+	+	_	+	+	+	+	_	+
NO ₂		_	_	_	_	+	+	_	_		_	_	_
OH			-	-	_	-	_	-	-	_	-	_	-
OH		_	-	_	_	-	_	-	_	-	-	-	_

				1000	Control	· · · · · · · · · · · · · · · · · · ·						
R	1	2	3	4	5 M	icrobial 6	Spectru 7	m ^a 8	9	10	11	12
OCH ₃	-	_	-	_	_	-	+	-	_	-	_	_
OCH ₃	-	_	_	_	+	-	~	+	_	_	_	_
CH ₃ O OCH ₃	_	_	_	-	-	_	~	_		_	_	
OC_2H_5	_	_	+	+	_	_	+	-	+	+	_	_
	_	_	_	_	_	_	+	+	+	+	_	+
$\Gamma_{\rm s}$	+	_	+	+	+	+	+	+	_	+	+	_

^a Microbial spectrum: Gram-positive—1, Staphylococcus aureus K257; 2, Mycobacterium smegnatis. Gram-negative—3, E. coli ATCC 4157; 4, Pseudomonas aeruginosa; 5, Klebsiella pneumoniae ATCC 8052; 6, Proteus vulgaris LBa 155; 7, Neisseria catarrhalis; 8, Saccharomyces sp.; 9, Candida albicans ATCC 10231; 10, S. epidermidis; 11, Aspergillus niger; 12, Trichophyton mentagrophytes ATCC 9129.

EXPERIMENTAL

Materials and Methods-The test organisms included Gram-positive Staphylococcus aureus K257, Mycobacterium smegmatis; Gram-negative E. coli ATCC 4157, Pseudomonas aeruginosa, Klebsiella pneumoniae ATCC 8052, Proteus vulgaris Lba 155, Neisseria catarrhalis; Saccharomyces sp., S. epidermidis, A. niger, Candida albicans ATCC 10231, and Trichophyton mentagrophytes ATCC 9129. The agar medium was inoculated heavily with the test organism and then filter paper disks (6.35 mm.) saturated with two drops of the solution of the test compound (20 mg./ml. in aqueous ethanol or water) were placed on the agar. After 48 hr. of incubation period, the zones of inhibition around the disks were measured. Zone sizes smaller than 6.35 mm. were considered minus activity.

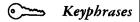
DISCUSSION

Seventeen β -aminoketones substituted at various positions in the aromatic ring were subjected to preliminary antibacterial and antifungal screening procedures. The substituents consisted of 4-fluoro,

4-chloro, 4-bromo, 4-nitro, 4-methyl, 4-phenyl, 4-hydroxy, 4-methoxy, 4-ethoxy, 3-nitro, 3-methoxy, 3-methyl, and 3-hydroxy. The activity against various organisms of these compounds is described in Table I. Hydroxy, fluoro, 4-methyl, 3,4,5-trimethoxy substituted β -amino ketones were completely inactive, whereas compounds with substituents such as chloro, bromo, and 4-nitro demonstrated the highest activity.

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 β -Amino ketones Antibacterial activity— β -amino ketones Antifungal activity— β -amino ketones Paper disk method—analysis